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A Synthesis of Emodin Anthrone

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Summary. A novel high yield (74% overall) synthesis of five steps to prepare emodin anthrone, which serves as a valuable precursor of hypericin, was devised using the benzamide ortho lithiation strategy for the key step.

Keywords. Emodin anthrone; Hypericin; Synthesis; Regioselective lithiation.

Eine Synthese von Emodinanthron

Zusammenfassung. Es wird eine neue Syntheseroute zur Darstellung von Emodinanthron, welches einen wichtigen Synthesevorläufer der Hypericinsynthese darstellt, beschrieben; sie verwendet als Schlüsselreaktion eine Benzamid-ortho-Lithiierungsreaktion und verläuft in fünf Stufen mit einer Gesamtausbeute von 74%.

Introduction

Emodin anthrone (1) serves as the immediate precursor in the course of hypericin synthesis $[1]$. A recent interest in hypericin (2), sparked by its antiviral and antiretroviral properties [2], spurred a search for efficient methods to improve its synthetic accessibility. According to literature, 2 may be either isolated from natural sources, as for instance *Hypericum perforatum* and *Hypericum hirsutum* [3], or it can be obtained by dimerization procedures starting from emodin derivatives [1, 4, 5]. However, accessibility using these roads suffers from rather long and low yield synthesis routes $\lceil 1, 4-6 \rceil$ or the tedious isolation from natural sources $\lceil 3, 6 \rceil$

^{*} Dedicated to Prof. K. Schaffner on the occasion of his 60th birthday

7]. We therefore developed a high yield synthetic route to prepare emodin anthrone as the key precursor of 2, which will be reported in this communication.

Results and Discussion

Lithiation of the readily accessible amide 3 (in two steps from the commercially available *meta-cresotinic* acid) following a recently described [8] strategy for benzamide $-ortho$ -metallation is observed to be strictly regioselective for position 6. This was proven by the nearly quantitative deuterium substitution of this position on treatment of the lithiation product of 3 with deuterium oxide. Reaction of the lithiation product of 3 with 3,5-dimethoxy-benzaldehyde and refluxing of the resulting crude carbinol 4 with p-toluenesulfonic acid in toluene afforded the lactone 5 in 94% overall yield. Upon hydrogenation of 5 using activated palladium on

Synthesis of Emodin Anthrone 741

charcoal as the catalyst the acid $\bf{6}$ was obtained in nearly quantitative yield. Cyclization of 6 by means of trifluoroacetic anhydride produced in good yield trimethyl emodin anthrone (7) which is exclusively present in solution as its 9-anthracenol tautomer. This was proven by the intensity and the chemical shift of its 1 H-NMR -H-10 signal, which are characteristic of *one aromatic* proton. The trifluoroacetyl derivative 8 could be isolated as a stable product of the acetylation of 7 by trifluoroacetic anhydride when the latter was used in a one molar excess. It was easily cleaved in high yield by hydrochloric acid in methanol to give 7, As 8 is found also as a by-product, diminishing the yield of the desired product 7 even when equimolar amounts of trifluoroacetic anhydride are used, it seemed to be preferable to execute the sequence from 6 via 8 to 7 as a one-pot procedure using a one molar excess of the trifluoroacetic anhydride. Demethylation of 7 by means of hydroiodic acid in acetic acid (in analogy to the preparation of emodin anthrone from trimethyl emodin [9]) produced 1 in a 90% yield. Thus, a reaction sequence was generated which affords emodin anthrone (1) starting from readily accessible 3 in five steps – the overall yield of this synthesis was 74% .

Experimental Part

Melting points were taken by means of a Kofler hot stage microscope (Reichert, Vienna). $H -$, ^{13}C -, IR-, UV-VIS-, and M-spectra were recorded using the Bruker-WM-360, and AC-200, Biorad-FT-IR-45, Hitachi-U-3210, and Finnigan-MAT-115 instruments. Tetrahydrofurane and benzene were dried by means of sodium benzophenone-their water contents were monitored by Karl Fischer titrations. *See-butyl* lithium solutions in n-pentane (Aldrich) were used after analysing their content by titration with *sec-butanol* in absol, benzene using 1,10-phenanthroline as the indicator.

$N, N\text{-}Diethyl-2-methoxy-4-methylbenzamide (3; C₁₃H₁₉NO₂)$

3 was prepared by methylation and amidation according to standard preparation methods of benzamides [10, 11] from 2-hydroxy-4-methyl-benzoic acid (Fulka) in 90% yield as white crystals of m.p. $58-61^{\circ}$ C and b.p. $128^{\circ}C/0.35$ mm. ¹H-NMR (CDCl₃, 200 MHz, δ): 7.06 (X-part of AMXsystem, $J_{AX} = J_{MX} = 8$ Hz, aryl-H 6), 6.77 (M-part of AMX-system, $J_{AX} = J_{MX} = 8$ Hz, $J_{AM} = 2$ Hz, aryl-H 5), 6.71 (A-part of AMX-system, J_{AM} = 2 Hz, aryl-H 3), 3.80 (s, CH₃O), 3.55 (q, broadened, $J=7$ Hz, CH₂CH₃), 3.14 (q, $J=7$ Hz, CH₂CH₃), 2.36 (s, CH₃), 1.23 (t, $J=7$ Hz, CH₂CH₃), 1.02 (t, $J=7$ Hz, CH₂CH₃) ppm. IR (KBr): 1 620, 1 500 cm⁻¹.

N,N-Diethyl-2-methoxy-4-methyl-6-[(3,5-dimethoxyphenyl)-hydroxymethyl]-benzamide $(4; C_{22}H_{29}NO_5)$

4 was prepared by regioselective lithiation of 3 in analogy to Ref. [10] and reaction with 3,5-dimethoxybenzaldehyde (Merck). Into a thoroughly flamed three-necked reaction flask 36 ml absol, tetrahydrofurane was introduced under argon atmosphere and cooled to -80° C. After introduction of 664 μ l (4.4 mmol) of tetramethylethylenediamine, 3.14 ml of a 1.4 molar *sec-butyl* lithium solution in pentane was added and the mixture was stirred for 10 min. To this intensively yellow colored solution of the complex a solution of 884mg (4mmol) of 3 in 4ml tetrahydrofurane was slowly added and the resultant mixture stirred for 20min. After fast addition of a solution of 731 mg (4.4mmol) 3,5 dimethoxy-benzaldehyde (Merck) in 4 ml absol, tetrahydrofurane still at - 80°C the reaction mixture was stirred for additional 20 min and afterwards quenched by addition of 2 ml water. The reaction mixture was acidified with 5% HC1, tetrahydrofurane evaporated in vacuo by means of a rotatory evaporator, and the resulting aqueous phase extracted three times with chloroform. Washings of the chloroform phase with satd. NaCl solution, drying with $Na₂SO₄$, and evaporation of solvent resulted in 1 550 mg of the practically pure raw product (100% yield) which can be used immediately in the following step. A yield of 1455mg (94%) of white crystals was obtained on crystallization from toluene; m.p. $142 - 143^{\circ}$ C. 1 H-NMR (CDCl₃, 360 MHz, δ): 6.90 and 6.67 (2 s, broadened, aryl-H 3, aryl-H 5), 6.52 (d, $J=2.2$ Hz, aryl-H 2', aryl-H 6'), 6.29 (t, 1 H, aryl-H 4'), 5.57 (d, $J=10$ Hz, CH-OH), 5.21 (d, $J=10$ Hz, CH-OH), 3.76 (s, CH₃O), 3.74 (s, 2CH₃O), 3.28 - 2.55 (m, broadened, $J=7.15$ Hz, CON(CH₂CH₃)₂), 2.38 (s, CH₃), 0.95 (t, $J=7.15$ Hz, CH₂CH₃), 0.85 (t, $J=7.15$ Hz, CH_2CH_3) ppm. ¹³C-NMR (CDCl₃, 50 MHz, δ): 168.7 (s, CONEt₂), 160.5 (s, 2C, C3', C5'), 155.9 (s, C6), 146.5, 143.0, 140.5 (3s, C1, C2, C4, CI'), 123.6, 111.0 (C3, C5), 104.0 (C2', C6'), 98.9 $(C4')$, 76.4 (CHOH), 55.3 (3 CH₃O), 43.0, 38.5 (2 CH₂CH₃), 21.7 (CH₃), 12.9, 12.4 (2 CH₂CH₃) ppm. The assignments were achieved using DEPT and ${}^{1}H-{}^{13}C-COSY$ experiments.

N,N-Diethyl-6-deutero-2-methoxy-4-methylbenzamide $(C_{13}H_{18}DNO_2)$

This compound was prepared according to the procedure described above to test for regioselective metallation by providing D₂O instead of the dimethoxybenzoic aldehyde as the electrophile. Yield 99% of white crystals of m.p. 60° C. ¹H-NMR (CDCl₃, 200 MHz, δ): 6.77 (A-part of AM system, J_{AM} =2Hz, aryl-H 5), 6.71 (M-part of AM system, J_{AM} =2Hz, aryl-H 3), 3.80 (s, CH₃O), 3.55 (q, broadened, $J = 7$ Hz, CH₂CH₃), 3.14 (q, $J = 7$ Hz, CH₂CH₃), 2.36 (s, CH₃), 1.23 (t, $J = 7$ Hz, CH₂CH₃), 1.02 (t, J=7Hz, CHzCH3) ppm. MS (25°C, 70 eV): *m/e* (%)=222 (12, M+), 220 (10), 151 (10), 150 (100), 92 (10).

$5-Methyl-7-methoxy-3-(3.5-dimethoxophenyl)-3H-isobenzofuranone (5; C₁₈H₁₈O₅)$

400 mg (1.03 mmol) crude 4 (as prepared above) and 30 mg p-toluenesulfonic acid were refluxed in 100 ml toluene (p. a.) for 2 h. The resulting solution was extracted twice with Na_2CO_3 (5% solution) followed by washings with water, HCl $(5%)$ and satd. NaCl solution. After drying over Na₂SO₄ the organic layer was evaporated and the residue crystallized from toluene. Yield 306 mg (94% overall yield from amide 3!) of white crystals of m.p. $160 - 161^{\circ}$ C. ¹H-NMR (CDCl₃, 200 MHz, δ): 6.73 (s, aryl-H 4 or H 6), 6.68 (s, aryl-H 4 or H 6), 6.43 (s, aryl-H 2', H 6', H 4'), 6.16 (s, H 3), 4.0 (s, CH₃O), 2.76 (s, 2CH₃O), 2.40 (s, CH₃) ppm. ¹³C-NMR (CDCl₃, 50 MHz, δ): 168.0 (CO), 161.2 (C3', C5'), 158.2 (C7), 152.7, 148.3, 139.2, 115.0 (C 1', C3a, C5, C7a), 111.9 (C4, C6), 104.5 (C2', C6'), 100.8 $(C4')$, 81.2 $(C3)$, 56.0 $(2 \text{CH}_3\text{O})$, 55.4 (CH_3O) , 22.5 (CH_3) ppm. The assignments were achieved using a 1H- 13C-COSY experiment. MS (90°C, 70 eV): *m/e* (%)= 316 (2, M+ 2), 315 (21, M+ 1), 314 (100, M^+), 285 (4), 284 (19), 241 (5), 177 (7), 149 (26), 148 (6), 142 (5). IR (KBr): 1 750, 1 610, 1 590 cm⁻¹.

2-Methoxy-4-methyl-6- $[(3,5\text{-}dimethoxyphenyl)-methyl]-benzoic Acid (6; C_{18}H_{20}O₅)$

800 mg lactone 5 (2.545 mmol) were dissolved in 50 ml absolute tetrahydrofurane + 50 μ l conc. H₂SO₄. After addition of 50 mg activated Pd/C (10%) the solution was agitated for 6h under a hydrogen atmosphere of 5 bar at 45°C. The activation of Pd/C was achieved by treatment with 10% sulfuric acid for 5 min and washings with water, methanol and absol, tetrahydrofurane. The filtered solution was evaporated and crystallized from toluene. 774 mg white crystals of m. p. $260 - 262$ °C were obtained (96% yield). ¹H-NMR (CDCl₃, 200 MHz, δ): 10.12 (s, broadened, COOH), 6.63 (s, aryl-H 5), 6.61 (s, aryl-H 3), 6.38 (d, $J=2.2$ Hz, aryl-H 2', H 6'), 6.29 (t, $J=2.2$ Hz, aryl-H 4'), 4.03 (s, CH₂), 3.85 (s, CH₃O), 2.72 (s, 2 CH₃O), 2.28 (s, CH₃) ppm. ¹³C-NMR (CDCl₃, 50 MHz, δ): 172.1 (COOH), 160.6 $(C3', C5')$, 157.0 $(C6)$, 142.6, 141.8, 140.7 $(C4, C2, C[†])$, 123.6 $(C3)$, 110.1 $(C5)$, 107.2 $(C2', C6')$, 98.2 (C 4'), 56.0 (CH₃O), 55.1 (2 CH₃O), 39.3 (CH₂), 21.7 (CH₃) ppm. The assignments were achieved using 1H- 13C-COSY and DEPT experiments. MS (70°C, 70 eV): *m/e* (%)= 316 (100, M+), 298 (12, M-H20), 283 (12), 149 (12), 99 (12), 97 (12), 85 (12), 83 (12), 77 (12), 73 (12), 71 (50), 69 (12), 60 (12), 59 (100), 57 (100), 55 (12). IR (KBr): 1 700, 1 600, 1 460 cm⁻¹.

Synthesis of Emodin Anthrone 743

1,3,8-Trimethoxy-6-methylanthracen-9-ol $(7; C_{18}H_{18}O_4)$

(a) 500 mg (1.27 mmol) of the trifluoroacetyl derivative 8 were dissolved in 600 ml methanol under an argon atmosphere. After addition of 20 μ l concentrated HCl the mixture was agitated at 45°C for 2 h. The solution was neutralized by means of satd. aqueous sodium carbonate, filtered off and evaporated. Crystallization from methanol resulted in 344 mg (91%) 7; yellow crystals, m.p. 161°C. ¹H-NMR (CDCl₃, 360 MHz, δ): 10.76 (s, 1 H, fast exchange upon addition of CD₃OD, OH), 7.48 (s, broadened, fast exchange upon addition of $CD₃OD$, H 10), 7.14 and 6.43 (2s, broadened, H 5 and H 7), 6.65 and 6.35 (AX system, $J_{AX} = 2.0$ Hz, H 2 and H 4), 4.05, 4.03, 3.91 (3s, 3 CH₃O), 2.44 $(s, CH₃)$ ppm. ¹³C-NMR (CDCl₃, 90 MHz, δ): 158.9, 157.7, 157.3, 153.4 (C 1, C 3, C 8, C 9), 135.8, 135.5 (C 4 a, C 10 a), 135.7 (C 6), 119.2 (C 7), 113.7 (C 5), 104.1 (C 4), 96.93, 96.90 (C 2, C 10), 109.4, 108.4 (C9 a, C 8 a), 56.13, 56.08, 55.34 (3 CH30), 22.04 (CH3) ppm. UV-VIS (ethanol): 372 (8 600), 265 (79 800), 231 (19 200) nm (e). MS (90°C, 70 eV): *role (%)=* 300 (20.9), 299 (47.6), 298 (100, M+), 283 (40), 255 (45), 240 (32), 225 (31), 197 (19), 149 (43). IR (KBr): 3300, 1620 cm⁻¹.

(b) 600 mg (1.896 mmol) benzoic acid 6 were dissolved in 70 ml absolute dichloromethane under an argon atmosphere. After addition of 528μ l trifluoroacetic anhydride (3.78 mmol) the mixture was agitated at room temperature for 6 h. Then the solvent was evaporated and procedure (a) as given above was followed to yield 514 mg $7 (91\% \text{ yield of the one-pot synthesis of } 7 \text{ from 6}).$

10- Trifluoroacetyl-1,3,8-trimethoxy-6-methylanthracen-9-ol (8; C₂₀H₁₇F₃O₄)

Following procedure (b) of the synthesis of 7, after evaporation of the solvent, the trifluoroacetic acid is removed at 0.01 torr at room temperature and the raw material is crystallized from absolute ethanol. From 600 mg acid 6 710 mg 8 (95%) are obtained as orange crystals; m.p. $174-176^{\circ}$ C. ¹H-NMR (CDCl₃, 360 MHz, δ): 11.27 (s, OH), 6.91 and 6.53 (2s, broadened, H 5 and H 7), 6.38 and 6.37 (2s, broadened, H 2 and H 4), 4.07 (s, CH₃O), 4.04 (s, CH₃O), 3.87 (s, CH₃O), 2.44 (s, CH₃) ppm. ¹³C-NMR (CDCl₃, 90 MHz, δ , 5% chromium acetylacetonate as relaxation reagent): 191.1 (q, J=35.5Hz, COCF3), 159.9, 159.6, 158.0. 157.9 (C1, C3, C8, C9), 139.2 (C6), 134.1, 133.5 (C4a, C10a), 115.2 (C7), 108.8, 107.6 (C9a, C8a), 104.9 (C5), 97.4 (C4), 93.5 (C2), 96.8 (C10), 56.4 $(2\text{CH}_3\text{O})$, 55.2 (CH₃O), 22.5 (CH₃) ppm; the CF₃ quartet could not be localized. UV-VIS (ethanol): 437 (7000), 323 (10 100), 265 (46 000), 223 (20 100) nm (ε). IR (KBr): 3 300, 1 620, 1 580 cm⁻¹. MS (90°C, 70eV): *m/e* (%)=396 (11), 395 (53), 394 (78, M+), 326 (60), 325 (100, M-CF3), 298 (58, M-COCF3), 297 (53), 282 (54), 197 (41).

Emodin Anthrone (1; $C_{15}H_{12}O_4$)

200 mg 7 (0.67 mmol) were dissolved in 18 ml acetic acid and 8 ml HI (ρ = 1.50) and the mixture kept under reflux for 4 h. Upon cooling the product crystallized from the reaction mixture. It was rinsed with water and dried [9]. 154 mg of 1 were obtained (90%); m. p. $254 - 258^{\circ}C$ (250 $- 258^{\circ}C$ [9]). ¹H-NMR (DMSO-d₆, 200 MHz, δ): 12.36 and 12.19 (2s, OH 1 and OH8), 10.82 (s, broadened, OH3), 6.73 and 6.64 (2s, broadened, H 5 and H 7), 6.37 and 6.20 (AM system, $J_{AM} = 2.2$ Hz, H 2 and H 4), 4.26 (s, CH2), 2.29 (s, CH3) ppm. 13C-NMR *(DMSO-d6,* 50MHz, 5): 191.0 (s, C=O), 164.9, 164.5, 161.6 (3s, C1, C3, C8), 146.9, 144.8, 141.9 (3s, C4a, C6, C10a), 112.7, 108.3 (2s, C9a, C8a), 119.8 (d, C7), 115.1 (d, C5), 107.3 (d, C4), 100.9 (d, C2), 32.2 (CH₂), 21.5 (CH₃) ppm. Assignments were achieved by means of a ${}^{1}H-{}^{13}C-COSY$ experiment. IR (KBr): 1620 cm⁻¹. UV (ethanol): 379 (18 300), 271 (15 300), 225 (26 200) nm(~). MS (90°C, 70eV): *m/e* (%)= 300 (21), 299 (48), 298 (100, M^+), 283 (40), 255 (45), 240 (32), 225 (31), 197 (19), 149 (43).

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